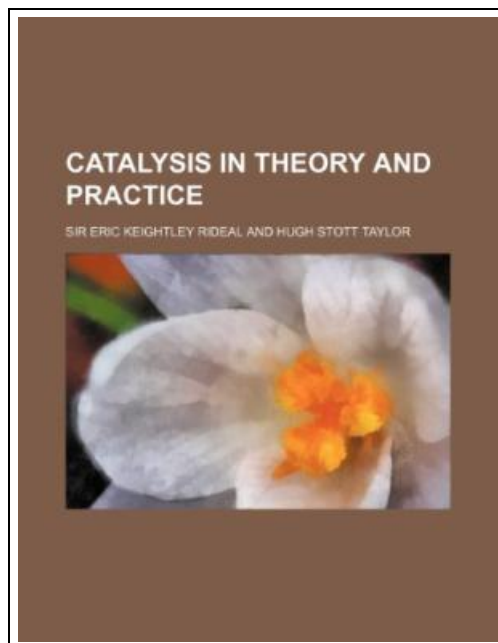


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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1919 Excerpt: .so slow as to render their technical operation a matter of some difficulty. The slowness of absorption in the formation of nitric acid is probably chiefly due to the slow reaction velocity of the above reaction and the need of some catalytic agency is clearly indicated. The use of small quantities of chlorine or bromine vapour has been suggested as efficient catalysts, but the difficulties incurred in elimination of the chlorine or bromine from the resulting nitric acid has been the chief objection to their extended use. THE PRODUCTION OF SYNTHETIC AMMONIA BY THE HABER PROCESS The technical production of pure ammonia by the catalytic combination of nitrogen and hydrogen must be considered as one of the greatest triumphs of modern physical and engineering chemistry. In 1865, Deville⁴ observed that the decomposition of ammonia under the influence of a spark discharge was never complete, thus indicating the reversibility of the reaction: $2\text{NH}_3 = \text{N}_2 + 3\text{H}_2$. The recent communication of Prof. C. Matignon to the 1 Zeitsch. Elektrochem., 1906, 12, 541. 2 Zeitsch. angew. Ohem., 1908, 21, 2161. Ibid., 1906, 19, 807. 4 Compt. rend., 1865, 60, 317. inaugural session of the Societe de Chimie Industrielle presents a very great interest to all concerned in ammonia synthesis in particular and also in the application of science to industry in general. The communication records a research into the scientific literature and more especially the patent literature concerning the production of ammonia from its elements. A resume of...

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