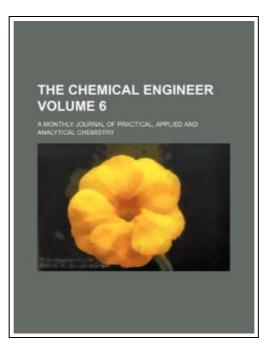
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RareBooksClub. Paperback. Book Condition: New. This item is printed on demand. Paperback. 112 pages. Dimensions: 9.7in. x 7.4in. x 0.2in.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1907 Excerpt: . . . slowly aspirated through the bromine in the absorbing bottles. When the aspirator bottle is empty, five liters of the producer gas have passed through the stopper of one of them. The apparatus is then taken back to the laboratory and the bottles are disconnected and their contents transferred to a 750 c. c. beaker, carefully washing out all of the bromine water with distilled water. Now add two grams of sodium carbonate to the solution and heat to boiling temperature, on the hot plate or other safe device. After the solution begins to boil add 20 c. c. of strong hydrochloric acid and continue boiling until the bromine is boiled off. The sulphur may now be precipitated by 20 c. c. of a hot ten per cent solution of barium chloride in water. The solution should be boiled vigorously for at least 30 minutes to insure the barium sulphate coming down in a form to facilitate easy filtration. A Gooch crucible is now fitted with an asbestos felt, ignited and weighed and the barium sulphate filtered through this, heated to a dull red over a Bunsen burner, cooled and weighed. The sulphur found is calculated for 100 cubic meters of gas at io C. and 760 mm. pressure. If 5. 0 liters have been used for the determination, and if denotes the temperature of...

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