

## Chemistry, Inorganic and Organic, with Experiments

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## Reviews

The book is great and fantastic. it had been writtern extremely perfectly and valuable. I am very happy to let you know that here is the finest pdf i have read through within my own life and can be he very best book for actually. (Miss Rossie Fay)

## CHEMISTRY, INORGANIC AND ORGANIC, WITH EXPERIMENTS



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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book \*\*\*\*\* Print on Demand \*\*\*\*\*.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1895 Excerpt: . easily imparts oxygen to other substances; sulphur, mixed with six times its weight of PbO., may be ignited by friction; hence this oxide is a common ingredient in lucifer-match compositions. Its oxidising property is frequently turned to account in the laboratory; for example, in absorbing sulphur dioxide from gaseous mixtures by converting it into sulphate of lead; PbO, + SO, = PbS04. Dioxide of lead is not dissolved by dilute acids, and has no basic properties, although certain salts of the type PbX4 are known; it is even sometimes called plumbic anhydride, for it acts upon potassium hydroxide, yielding potassium plumbate (KjPbOjHjO), which has been crystallised from an alkaline solution, but is decomposed by pure water. Lead dioxide evolves 01 from HC1 when heated, and gives, at first, a brown solution (containing PbCl4) which yields a brown precipitate with ammonia, but if the solution be boiled till all the Cl is expelled, it becomes colourless PbCl, and gives a white precipitate with ammonia. PbO is converted into PbO, by ozone and by hydrogen peroxide. Lead hydroxide, Pb(OH), has not been obtained, but Pb(OH)s.PbO is formed as a white precipitate when air and water, free from CO, attack lead. The same substance is precipitated by alkalies from solutions of lead salts. It becomes PbO when heated to 145 C. The compound Pb(OH), .2PbO crystallises in octahedra from a solution of basic lead acetate mixed with ammonia. Lead nitrate, Pb(NO3), crystallises in white octahedra from a solution...

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